



RESEARCH & ANALYTICAL LABORATORIES, INC.

Analytical/Process Consultations

07 April 2011

Compliant Service Branch
Public Water Supply Section
1634 Mail Service Center
Raleigh, NC 27699-1634
Attn: Christyn Fertenbaugh

Re: Corrosion Control Evaluation
PWSID No. NC0241410
Brown Summit Middle School
Guilford County Schools

Dear Ms. Fertenbaugh:

Enclosed please find the Corrosion Control Treatment Study (CCT) for the above referenced facility. This groundwater system has exceeded the copper action level (i.e. 2.97 mg/l –October 2007). It is important to note that this water supply system has been in lead and copper compliance for the last four (4) years and should be eligible for reduced monitoring. It is suspected that inadequate flushing may have been the cause of the 2007 copper level exceedance. It is therefore recommended that the facility apply for reduced monitoring, carefully monitored (i.e. pH, copper, etc.), and no further action be taken until the lead or copper exceedance can be documented. These recommendations are based upon current and historical lead and copper data, water quality parameters, and the US EPA Revised Guidance Manual for Selecting Lead and Copper Control Strategies.

If you should have any questions concerning this report, or need additional information please so advise.

Sincerely,

James M. Cheshire
President/CEO
Research & Analytical Labs, Inc.

APR 19 2011

JMC/js



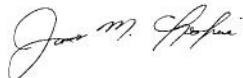
Evaluation Form for Corrosion Control Treatment

MAIL TO:

Compliance Branch
Public Water Supply Section
1634 Mail Service Center
Raleigh, North Carolina 27699-1634

A. PWS General InformationDate: 04/05/2011

1. PWS Name: Brown Summit Middle School
2. PWSID Number: NC0241410
3. Contact Person:
Name: Mr. Larry Odell
Mailing Address: Brown Summit Middle School
3920 Naco Road
Greensboro, North Carolina 27405
Telephone: 336-370-2387
Fax: 336-370-2398
4. Population served: 236
5. Person Responsible for preparing this form:

Name: James M. CheshireSignature: Telephone: 336-996-2841**PWS Technical Information**

1. Monitoring Results:

Sampling Dates: From 07/01/07 To 12/31/07

First-Flush Tap Monitoring Results:

Lead:

Minimum concentration =	<u><0.003</u> mg/L
Maximum concentration =	<u><0.003</u> mg/L
90th percentile =	<u><0.003</u> mg/L

Copper:

Minimum concentration =	<u>1.24</u> mg/L
Maximum concentration =	<u>4.05</u> mg/L
90th percentile =	<u>2.97</u> mg/L

Point of Entry Monitoring Results (Averages): Water Quality Parameters (March 28 & 29 2011)

	Points of Entry				
	1	2	3	4	5
Lead Concentration in mg/L:	<u><RRL</u>	_____	_____	_____	_____
Copper Concentration in mg/L:	<u><RRL</u>	_____	_____	_____	_____
pH:	<u>6.3</u>	_____	_____	_____	_____
Temperature, °C:	<u>12</u>	_____	_____	_____	_____
Alkalinity, mg/L as CaCO ₃ :	<u>20</u>	_____	_____	_____	_____
Calcium, mg/L as Ca:	<u>2.84</u>	_____	_____	_____	_____
Conductivity, μ mho/cm @ 25° C:	<u>83</u>	_____	_____	_____	_____
Phosphate, mg/L as P:	<u><RRL</u>	_____	_____	_____	_____
Silicate, mg/L as SiO ₂ :	<u>7.53</u>	_____	_____	_____	_____

*DIC = 9 mg C/L (saturation pH = 9.5)

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Water Quality Parameter Distribution System Monitoring Results:
Indicate whether field or laboratory measurement.

	Field	Lab
pH: minimum = <u>6.3</u> maximum = <u>6.5</u>	X	_____
Alkalinity: minimum = <u>19</u> mg/L as CaCO ₃ maximum = <u>20</u> mg/L as CaCO ₃	_____	X
Temperature: minimum = <u>13</u> °C maximum = <u>18</u> °C	X	_____
Calcium: minimum = <u>2.80</u> mg/L as Ca maximum = <u>2.87</u> mg/L as Ca	_____	X
Conductivity: minimum = <u>83</u> μ mho/cm @ 25° C maximum = <u>84</u> μ mho/cm @ 25° C	_____	X
Orthophosphate: (if phosphate based inhibitor is used) minimum = <u><RRL</u> mg/L as P maximum = <u><RRL</u> mg/L as P	_____	X
Silica: (if silica based inhibitor is used) minimum = <u>7.49</u> mg/L as SiO ₂ maximum = <u>7.52</u> mg/L as SiO ₂	_____	X

2. Existing Conditions:

Is treatment used? yes no

Identify water source(s):

Source No. 1 P01 (Well)

Source No. 2 _____

Source No. 3 _____

If treatment is used, is more than one source used at a time?

 yes no

Identify treatment processes used for each source:

Process	No. 1	No. 2	No. 3
Presedimentation	_____	_____	_____
Aeration	_____	_____	_____
Chemical mixing	_____	_____	_____
Flocculation	_____	_____	_____
Sedimentation	_____	_____	_____
Recarbonation	_____	_____	_____

Identify treatment processes used for each source:

Process	No. 1	No. 2	No. 3
2nd Stage mixing	_____	_____	_____
2nd Stage flocculation	_____	_____	_____
2nd Stage sedimentation	_____	_____	_____
Filtration:	_____	_____	_____
Single medium	_____	_____	_____
Dual media	_____	_____	_____
Multi-media	_____	_____	_____
GAC cap on filters	_____	_____	_____
Disinfection:	_____	_____	_____
Chlorine	X	_____	_____
Chlorine dioxide	_____	_____	_____
Chloramines	_____	_____	_____
Ozone	_____	_____	_____
Granular Activated Carbon	_____	_____	_____

List chemicals normally fed:

List chemicals sometimes fed:

3. Present Corrosion Control Treatment:

None X

Inhibitor _____

Date initiated _____

Present dose _____ mg/L

Range in Residual in Distribution System:

Maximum _____ mg/L Minimum _____ mg/L

Brand name _____

Type _____

Has it been effective? Please comment on your experience.

pH/alkalinity adjustment _____

pH Target _____

Alkalinity Target _____ mg/L CaCO₃

Calcium adjustment _____

Calcium Target _____ mg/L CaCO₃

4. Water Quality:

Complete the table below for typical untreated and treated water quality data. Copy this form as necessary for additional sources. Include data for each raw water source, if surface supplies are used, and finished water quality information (point of entry) from each treatment plant. If wells are used, water quality information from each well is acceptable but not necessary if several wells have similar data. For groundwater supplies, include a water quality summary for each wellfield or grouping of wells with similar quality.

Include available data for the following:

Parameters	Untreated Supply	Treated Water (point of entry)
pH, units	N/A	6.29, 6.35
Alkalinity, mg/L as CaCO ₃	N/A	19.0, 21.8
Conductivity, μmho/cm @ 25° C	N/A	79.7, 86.5
Total dissolved solids, mg/L	--	--
Calcium, mg/L Ca	N/A	2.89, 2.78
Hardness, mg/L as CaCO ₃	N/A	7.23*, 6.95*
Temperature, °C	N/A	10.3, 12.8
Chloride, mg/L	--	--
Sulfate, mg/L	--	--

*Calculated

5. Distribution System:

Does the distribution system contain lead service lines?

yes no

If your system has lead service lines, mark below the approximate number of lines which can be located from existing records.

None _____ Some _____ Most _____ All _____

Is the distribution system flushed?

None Some Most All

6. Historical Information:

Is there a history of water quality complaints?

yes no

If yes, then answer the following:

Are the complaints documented? yes no

Mark the general category of complaints below. Use:

- 1 for some complaints in this category
- 2 for several complaints in this category
- 3 for severe complaints in this category

Categories of complaints:

Taste and odor _____

Color _____

Sediment _____

Other (specify) _____

Have there been any corrosion control studies?

yes no

If yes, please indicate:

Date(s) of study From _____ To _____

Study conducted by PWS personnel? yes no

Brief results of study were _____

Study results attached? yes no

Were treatment changes recommended? yes no

If yes:

Were treatment changes implemented? yes no

Have corrosion characteristics of the treated water changed? yes no

If yes, how has change been measured?

General observation _____

Coupons _____

Frequency of complaints _____

Other _____ Briefly indicate below:

7. Treatment Constraints:

Optimal corrosion control treatment means the corrosion control treatment that minimizes the lead and copper concentrations at users' taps while insuring that the treatment does not cause the water system to violate any State or national primary drinking water regulations. Please indicate below which constraints to treatment will apply to your PWS. Use the following code:

- 1 Some constraint = Potential Impact but Extent is Uncertain.
- 2 Significant constraint = Other Treatment Modifications Required to Operate Option.
- 3 Severe constraint = Additional Capital Improvements Required to Operate Option.
- 4 Very severe constraint = Renders Option Infeasible.

Constraint	Treatments		
	pH/Alkalinity adjustment	Calcium adjustment	Inhibitor PO ₄ Si
A. Regulatory	1	1	1 1
SOCs/OCs	1	1	1 1
SWTR: Turbidity	1	1	1 1
Total Coliforms	1	1	1 1
SWTR/GWDR: Disinfection	1	1	1 1
Disinfection Byproducts	1	1	1 1
Lead and Copper Rule	1	1	1 1
Radionuclides	1	1	1 1
B. Functional	1	1	1 1
Taste & Odor	1	1	1 1
Wastewater Permit	1	1	1 1
Aesthetics	1	1	1 1
Operational	1	1	1 1
Other	1	1	1 1

8. Evaluation:

Briefly summarize the review of the corrosion control literature that pertains to your PWS. A report or summary can be appended to this form if preferred.

N/A

Were other similar facilities located which are experiencing successful corrosion control? N/A
 yes no

If yes, identify their corrosion control treatment method.

None	_____
pH/Alkalinity adjustment	_____
Calcium adjustment	_____
Inhibitor	_____
Phosphate based	_____
Silica based	_____

9. Recommendations:

The corrosion control treatment method installed or being proposed is:

pH/Alkalinity adjustment
 Target pH is 8.5 units
 Target alkalinity is 80 mg/L as CaCO₃

Calcium adjustment
 Target calcium concentration is 40 mg/L Ca

Inhibitor N/A
 Phosphate based _____
 Brand name _____
 Target dose _____ mg/L
 Target residual _____ mg/L orthophosphate as P
 Silica based _____
 Brand name _____
 Target dose _____ mg/L
 Target residual _____ mg/L as SiO₂

Rationale for the proposed corrosion control treatment is:*

Discussed in the enclosed report

[EPA Revised Guidance Manual for Selecting Lead & Copper Control Strategies](#)

Briefly explained below

*Non-compliance data is based upon 2007 lead and copper monitoring data which had not been issued a NOV by DENR until now (2011). This system has been in consistent compliance for the last four (4) years and should be eligible for reduced monitoring. It is suspected that inadequate flushing of the distribution lines may have been the cause of the 2007 non-compliance. It is recommended that the system apply for reduced monitoring, carefully monitored (i.e. pH, etc.) and no further action be taken until a lead and copper exceedance can be documented.

List your proposed or operating guidelines:

Parameter	Operating Range
pH	7.0 – 8.5
Alkalinity	50 – 75 mg/L
Hardness	30 – 50 mg/L
_____	_____
_____	_____
_____	_____
_____	_____

Briefly explain why these guidelines were selected.

To improve the buffering capacity and pH stabilization of source water should raw water quality deteriorate.

10. Please provide any additional comments that will assist in determining optimal corrosion control treatment for your PWS.

See explanation in section 9. Note that no further action is recommended other than applying for reduced monitoring.